



## Cespitulins A–D, novel diterpenoids from Taiwanese *Cespitularia taeniata*

Yu-Chi Lin, Ahmed Eid Fazary, Ya-Ching Shen\*

School of Pharmacy, College of Medicine, National Taiwan University, Jen-Ai Rd. Sec. 1, Taipei 100, Taiwan, ROC

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### ABSTRACT

Four novel diterpenoids, designated cespitulins A–D (**1–4**) were isolated and characterized from the Taiwanese soft coral *Cespitularia taeniata*. Their structures were determined by extensive spectroscopic analyses ( $^1\text{H}$ – $^1\text{H}$  COSY, HMQC, HMBC, NOESY). Compounds **1–4** possess an unprecedented bond cleavage between C-9 and C-10 with a hemiacetalic lactone ring rather than a verticillane skeleton.

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### 1. Introduction

Soft corals are widespread in shallow waters at the southern coast of Taiwan. Species of the genus *Cespitularia* are characterized by polyps on their soft branches with white, cream, blue, brown, or iridescent–green surfaces.<sup>1</sup> It is a rich source of secondary metabolites with novel, diverse chemical frameworks, and interesting biological activities such as antitumor, antibacterial, and immunomodulatory activities.<sup>2–4</sup> Soft corals belonging to *Cespitularia* were reported to produce various diterpenoids possessing sesquiterpene, caryophyllane, neodolabellane, dolabellane, cespitularane, cespitulactone, cespitulactam, norverticillane, and verticillane skeletons.<sup>5–12</sup> Due to carbon skeletons of these secondary metabolites being very close to the taxane diterpenoids isolated from terrestrial yew,<sup>13,14</sup> it is very interesting to investigate their structures and structure activity relationships of verticillane analogues.

Our previous chemical study of this species has resulted in the isolation of a series of verticillene diterpenes. Some of these compounds are nitrogen-containing molecules.<sup>15–20</sup> In our constant research programs oriented toward the investigation for bioactive natural products from Taiwanese soft corals of the genus *Cespitularia*, the EtOAc/CH<sub>2</sub>Cl<sub>2</sub> extract of *C. taeniata* was investigated. This Letter reports the chemical investigation of new secondary metabolites from *C. taeniata*. The examination of different chromatographic fractions of *C. taeniata* extract has led to the isolation of four novel diterpenoids, designated cespitulins A–D (**1–4**) (Fig. 1). The structures of these compounds were elucidated through detailed spectroscopic analyses, mainly 2D NMR experiments ( $^1\text{H}$ – $^1\text{H}$ -COSY, HMQC, HMBC). The configuration at the chiral centers and the geometry of the double bonds were deduced from NOESY spectra and by application of molecular modeling.

### 2. Results and discussions

Compounds **1** and **2** were obtained as light yellow amorphous mixture, and both of them had a molecular formula C<sub>20</sub>H<sub>30</sub>O<sub>5</sub> as derived from its HR-ESI-MS and NMR spectroscopic data, indicating six degrees of unsaturation. The IR spectrum of compounds **1** and **2** suggested the presence of a diagnostic hydroxy (3385 cm<sup>-1</sup>) and lactone (1738 cm<sup>-1</sup>) groups. The  $^1\text{H}$  NMR spectroscopic features of **1** is similar to that of **2** (Table 1), including a doublet at  $\delta$  5.44 ( $J = 7.8$  Hz, H-7), a doublet of doublet at  $\delta$  4.52 ( $J = 14.4$ , 7.5 Hz, H-6), a pair of singlet at  $\delta$  4.91 and 4.93 in addition to three methyl singlets ( $\delta$  1.05, 1.26, 1.75). The comparison of  $^{13}\text{C}$  NMR spectrum for both **1** and **2** displayed signal of an ester carbonyl ( $\delta_{\text{C}}$  170.0 s for **1** and **2**) two trisubstituted olefinic carbons ( $\delta_{\text{C}}$  137.7, 127.0, and  $\delta_{\text{C}}$  137.7, 127.1 for **1** and **2**, respectively), an exocyclic double bond ( $\delta_{\text{C}}$  145.6 s, 113.3 t, and  $\delta_{\text{C}}$  145.5 s, 113.3 t for **1** and **2**, respectively), and three corresponding methyl carbons ( $\delta_{\text{C}}$  20.7 q, 24.3 q, 14.0 q, and  $\delta_{\text{C}}$  20.3 q, 23.8 q, 14.0 q for **1** and **2**, respectively). Subsequently, the methylene of H-9 shifted downfield to  $\delta_{\text{H}}$  4.00 suggesting that a hydroxyl group is attached at C-9. Besides, a pair of tetrasubstituted olefinic carbons appeared at  $\delta_{\text{C}}$  136.4 and 158.6 for **1** and  $\delta_{\text{C}}$  136.4 and 158.9 for **2** were characterized. The assignment of compounds **1** and **2** was completely established by two-dimensional NMR experiments, in which the  $^1\text{H}$ – $^1\text{H}$  COSY experiment showed two sets of correlations (H-13/H-14, H-14/H-1, H-1/H-2, H-2/H-3 as well as the H-5/H-6, H-6/H-7) and these

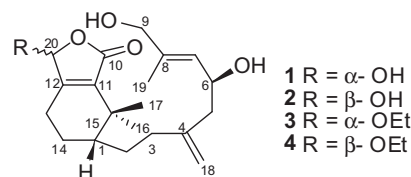
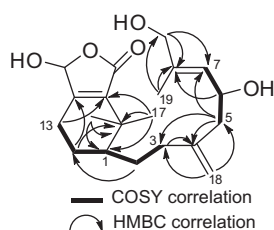


Figure 1. Compounds **1–4** isolated from the soft coral *Cespitularia taeniata*.

\* Corresponding author. Tel.: +886 2 23123456x62226; fax: +886 2 2391 9098.  
E-mail address: [ycshen@ntu.edu.tw](mailto:ycshen@ntu.edu.tw) (Y.-C. Shen).

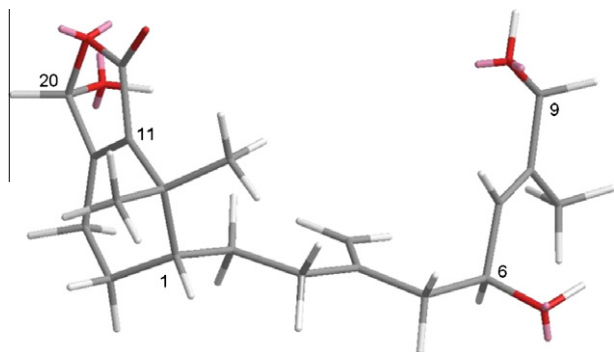
**Table 1**<sup>1</sup>H NMR spectroscopic data of **1–4** (mult, *J* in Hz)

No.	<b>1</b> <sup>a</sup>	<b>2</b> <sup>a</sup>	<b>3</b> <sup>a</sup>	<b>4</b> <sup>b</sup>
1	1.30 (m)	1.30 (m)	1.22 (m)	1.28 (m)
2	1.17, 1.67 (m)	1.17, 1.67 (m)	1.10, 1.70 (m)	1.19 (m), 1.71 (m)
3	2.00, 2.20 (m)	2.00, 2.20 (m)	1.98, 2.20 (m)	2.04 (dd, 8.5, 15.5)
5	2.20 (m)	2.20 (m)	2.20 (m)	2.28 (dm, 4.5)
6	4.52 (dd, 7.6, 14.8)	4.52 (dd, 7.6, 14.8)	4.50 (m)	2.25 (d, 7.0)
7	5.45 (d, 8.4)	5.45 (d, 8.4)	5.44 (d, 8.8)	4.55 (dd, 8.5, 15.0)
9	4.00 (s)	4.00 (s)	3.99 (s)	5.49 (dd, 1.5, 8.5)
12				4.04 (s)
13	2.10, 2.45 (m)	2.10, 2.45 (m)	2.20, 2.28 (m)	2.42 (m)
14	1.37, 1.95 (m)	1.37, 1.95 (m)	1.33, 1.85 (m)	2.12 (dd, 5.0, 19.0)
16	1.26 (s)	1.26 (s)	1.27 (s)	1.38 (m)
17	1.00 (s)	1.02 (s)	1.00 (s)	1.90 (dm, 6.0)
18	4.87 (s)	4.90 (s)	4.85, 4.88 (s)	1.30 (s)
19	1.70 (s)	1.70 (s)	1.70 (s)	1.05 (s)
20	5.81 (s)	5.83 (s)	5.53 (s)	4.91, 4.93 (s)
21			3.68, 3.87 (m)	1.74 (s)
22			1.22 (t, 6.0)	5.57 (s)
				3.72, 3.90 (dq, 9.5, 7.5)
				1.28 (t, 7.5)

<sup>a</sup> Measured at 400 MHz.<sup>b</sup> Measured at 500 MHz.**Figure 2.** COSY and HMBC correlations of **1**.

two sets of proton sequences were further connected by the HMBC correlations of H-18/C-3, C-4, C-5 as shown in Figure 2. Furthermore, the HMBC correlations of Me-16/C-1, C-15, C-11, and H-14/C-15, C-12 (Fig. 2) clearly indicated that **1** possesses an 2',2'-dimethyl cyclohexene moiety. The extraordinary bond cleavage between C-9 and C-10 was deduced from HMBC experiment, which clearly showed correlations of H<sub>2</sub>-9/C-7, C-8, C-19 but lacked for correlations of H<sub>2</sub>-9/C-10 (Fig. 2). In addition, compounds **1** and **2** possessed a hemiacetalic lactone ring system by virtue of HMBC correlations from H-20 to C-11, C-12, and C-10.<sup>12</sup> Thus, the gross structures of compounds **1** and **2** were determined.

A computer generated 3D chemical model for **1** by using MM2 force field calculation is illustrated in Figure 3, clarifying that com-

**Figure 3.** Computer-generated perspective model for **1** using MM2 force field calculation.

pounds **1** and **2** have three asymmetric carbon centers (C-1, C-6, C-20). Based on previously published cespiphytins and biogenetic consideration, the configuration of the hydroxy at C-6 was tentatively assigned at β.<sup>4,20</sup> In the course of our investigation, **1** and **2** were in 1:1 amount mixture by comparing the intensity of <sup>13</sup>C-NMR data.

The difference appeared on C-20 ( $\delta_c$  96.6, 92.2, respectively) indicating that it was obtained as an inseparable epimeric mixture due to the hemiacetalic inversion. Based on the above detailed explanation, the gross structures of both **1** and **2** are assumed to be as shown in Figure 1 and designated as cespitulins A (**1**) and B (**2**).

In the less polar fraction, cespitulin C (**3**) was isolated as an amorphous yellow powder,  $[\alpha]_D^{22} +28.3$  ( $c = 0.25$ , CH<sub>2</sub>Cl<sub>2</sub>), which had the molecular formula C<sub>22</sub>H<sub>34</sub>O<sub>5</sub> as derived from HR-ESI-MS,

**Table 2**<sup>13</sup>C NMR spectroscopic data of **1–4**

No.	$\delta_c$			
	<b>1</b> <sup>a</sup>	<b>2</b> <sup>a</sup>	<b>3</b> <sup>a</sup>	<b>4</b> <sup>b</sup>
1	44.1 d	44.3 d	44.1 d	44.5 d
2	26.5 t	26.4 t	26.4 t	26.5 t
3	34.2 t	34.4 t	33.8 t	34.3 t
4	145.6 s	145.5 s	145.6 s	145.6 s
5	43.9 t	43.9 t	43.9 t	44.0 t
6	65.8 d	65.8 d	65.7 d	65.7 d
7	127.1 d	127.0 d	127.1 d	127.2 d
8	137.7 s	137.7 s	137.6 s	137.6 s
9	67.7 t	67.8 t	67.8 t	67.9 t
10	170.0 s	170.0 s	169.8 s	170.0 s
11	136.4 s	136.4 s	136.8 s	136.9 s
12	158.6 s	158.9 s	157.0 s	157.4 s
13	22.7 t	22.4 t	22.7 t	22.6 t
14	22.8 t	22.9 t	22.9 t	22.9 t
15	33.8 s	33.7 s	34.2 s	33.8 s
16	24.3 q	23.8 q	24.2 q	23.6 q
17	20.7 q	20.3 q	20.3 q	20.6 q
18	113.3 t	113.3 t	113.3 t	113.3 t
19	14.0 q	14.0 q	14.0 q	14.0 q
20	96.6 d	92.2 d	101.5 d	101.1 d
21			65.8 t	65.7 t
22			15.0 q	15.0 q

<sup>a</sup> Measured at 100 MHz.<sup>b</sup> Measured at 125 MHz.

indicating six degrees of unsaturation. The IR spectrum revealed that this compound has a hydroxyl ( $3444\text{ cm}^{-1}$ ) and a lactone ( $1731\text{ cm}^{-1}$ ) group. The  $^1\text{H}$ - and  $^{13}\text{C}$  NMR spectra of **3** (Tables 1 and 2) were superimposable to those of **2** except that **3** contains signals of *O*-ethyl protons at  $\delta_{\text{H}}$  3.68 m, 3.87 m and 1.22 t ( $J = 6.0\text{ Hz}$ ) and carbons at  $\delta_{\text{C}}$  65.8 t and 15.0 q.

However, the acetalic proton shifted upfield to  $\delta_{\text{H}}$  5.53s (H-20) in **3** and its corresponding carbon shifted downfield to  $\delta_{\text{C}}$  101.5. The  $^1\text{H}$ - $^1\text{H}$  COSY of **3** determine three sets of correlations (Fig. 4). Moreover, the HMBC of **3** revealed correlations between H-21 and C-20.

Furthermore, the relative stereochemistry of **3** was determined by the NOESY correlations (Fig. 5). Assuming that **3** was  $\beta$ -configuration of H-1 the same with cespiphyptins,<sup>4,20</sup> NOESY experiments would assign the configuration of H-20. The presence of mutual correlations between H-1/H-13 $\beta$ , H-14 $\beta$ , Me-17; H-14 $\alpha$ /H-13 $\alpha$ , H-20 and the absence of correlation between H-20 and H-13 $\beta$  suggested that the configuration of H-20 is  $\alpha$ -oriented. On the other hand, the absence of correlation between H-7 and Me-19 suggested the *E* geometry of C-7 and C-8.

Compound **4** was isolated as a light yellow powder. Its IR,  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectroscopic data (Tables 1 and 2) are almost same as those of **3**, while its physical property is different in some manners. Compound **4** has a specific rotation  $[\alpha]_{\text{D}}^{25} -36.8$  (c 0.25,  $\text{CH}_2\text{Cl}_2$ ). On the HPLC separations, using RP-C<sub>18</sub>, (MeOH/ $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 5:1:4, flow: 2 ml/min), compounds **3** and **4** were separated at two different retention times, 42.6 and 46.6 mins, respectively. Also, some differences were shown by comparing the HMBC correlations of **4** [H-18/C-3, C-5, H-3, H-5/C-4, Me-19/C-7, C-8, C-9, H-9/C-7, C-8, H-20/C-10, C-11, C-12, C-21, H-13/C-11, C-12, H-21/C-20] (Fig. 6) with those of **3**.

The establishment of the relative three-dimensional structure was mainly deduced by virtue of the comparison with **3**. Some significant differences were observed in the NOESY spectrum of **4** (Fig. 7). The cross peaks between H-1/Me-17, H-13 $\beta$ , H-14 $\beta$ , and H-20/H-13 $\beta$ , H-13 $\alpha$  confirmed the proton assignment between C-13 and C-14, therefore, H-20 was in the  $\beta$  position.

A molecular model of structure **4** was generated by CS Chem 3D version 9.0 using MM2 force field calculation for energy minimiza-

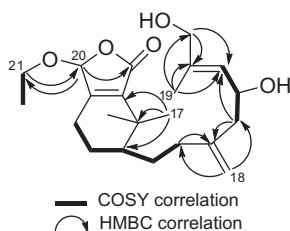


Figure 4. COSY and HMBC correlations of **3**.

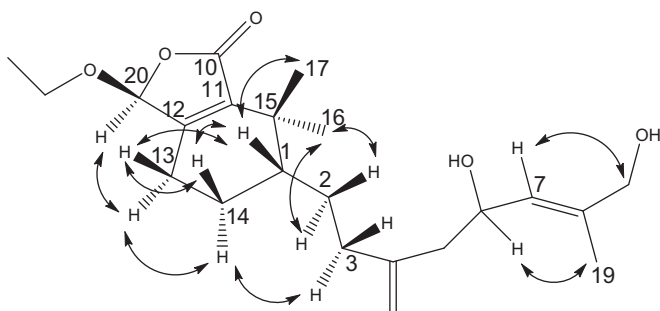


Figure 5. Selective NOESY correlations of **3**.

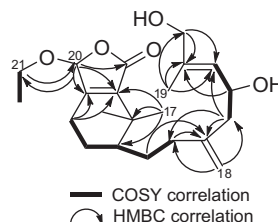


Figure 6. COSY and HMBC correlations of **4**.

tion (ChemBioUltra Calculation program) as shown in Figure 7. The result was consistent with the NOESY experiment. Based on the above interpretation, the structure of **4** was shown in Figure 1, and the name cespitulins D was given.

Compounds **1–4** represent new verticilene-like diterpenoids having an unprecedented bond cleavage between C-9 and C-10 with a hemiacetalic lactone ring rather than a verticillane skeleton. However, these compounds were inactive as tested for their in vitro cytotoxicity against human tumor cells. Compounds **1** and **2** may be biogenetically derived from cespiphyptin V<sup>20</sup> through steps of dehydration, oxidation leading to double bond cleavage, and reduction of ketone and aldehyde groups to **1** and **2** as illustrated in Scheme 1.

### 3. Experimental

#### 3.1. General

Optical rotations were recorded on a JASCO DIP-1000 polarimeter. IR spectra were recorded using a Horiba FT-720 spectrophotometer. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra as well as 2D NMR spectra (COSY, HMQC, HMBC, and NOESY) were recorded in  $\text{CDCl}_3$  (or  $\text{CD}_3\text{OD}$ ) using Bruker DRX NMR spectrometers operating at 300 or 500 MHz for  $^1\text{H}$  and 75 or 125 MHz for  $^{13}\text{C}$  using the  $\text{CDCl}_3$  solvent peak as internal standard ( $\delta$  7.26 for  $^1\text{H}$  and  $\delta$  77.0 for  $^{13}\text{C}$ ). Low-resolution EIMS spectra were recorded on a VG Quattro 5022 mass spectrometer. High-resolution ESIMS spectra were measured on a JEOL HX 110 mass spectrometer. Silica gel 60 (Merck) and Sephadex LH-20 (Amersham Pharmacia Biotech AB, Sweden) were used for column chromatography. Lichrosorb Si-60 (Merck) and Lichrosorb RP-18 (Merck) were used for HPLC column.

#### 3.2. Animal material

*Cespitularia taeniata* May was collected in Green Island, Taiwan, in March 2004. This soft coral was identified by one of the authors (YCS). A voucher specimen (GSC-1) was deposited in the School of Pharmacy, National Taiwan University, Taipei, Taiwan.

#### 3.3. Extraction and isolation

The soft coral (1.1 kg) was freeze-dried, powdered, and extracted with mixture of  $\text{CH}_2\text{Cl}_2/\text{EtOH}$  (1:1), and the crude extract was partitioned between EtOAc and  $\text{H}_2\text{O}$  (1:1). The EtOAc-soluble fraction (100 g) was subjected to a Si gel column (*n*-hexane/EtOAc, 15:1–0:1; EtOAc/MeOH, 50:1–2:1) to give fractions 1–12. Fraction 10 (540 mg) was separated with a Sephadex LH-20 column (MeOH) to yield a residue (400 mg), which then chromatographed extensively on a Si gel column (*n*-hexane/EtOAc, 3:1), LH-20 Sephadex resin ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 1:1) and HPLC (Si gel, *n*-hexane-acetone, 4:1; RP-C<sub>18</sub>, MeOH/ $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 6:1:3) to furnish a mixture residue, cespitulins A and B (**1** and **2**, 1 mg), cespitulactam C (12 mg),<sup>7</sup> and cespitularin F (95 mg).<sup>6</sup> Fraction 9 (2.5 g) was separated with a LH-20 Sephadex resin ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 1:1), Si gel column and HPLC (Si gel, *n*-hexane-acetone, 4:1; RP-C<sub>18</sub>, MeOH/

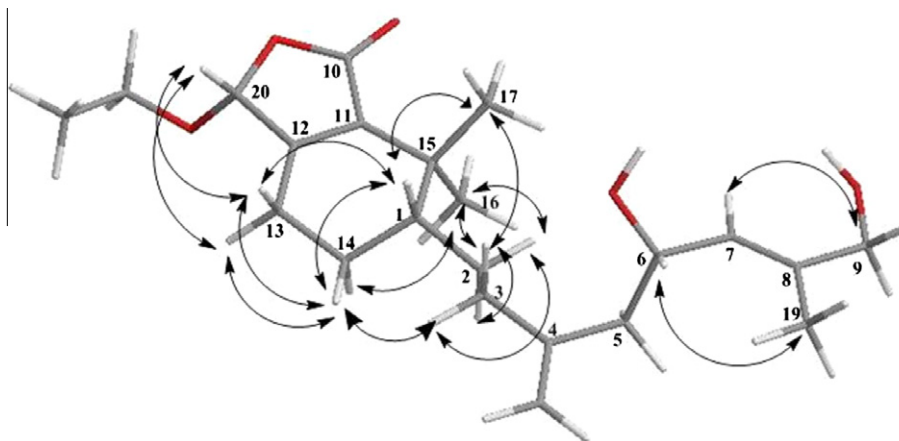
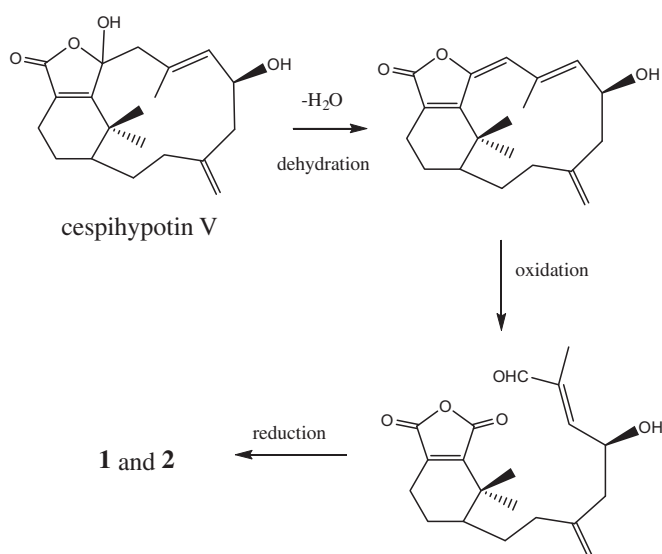


Figure 7. NOESY correlations of 4.



Scheme 1. Plausible biogenetic pathway to compounds 1 and 2.

$\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 5:1:4) to furnish cespitulins C (**3**, 5 mg), cespitulin D (**4**, 5 mg), and cespitulactam A (11 mg).<sup>7</sup>

Cespitulins A and B (**1** and **2**): light yellow powder.  $[\alpha]_{\text{D}}^{25} +15.8$  (c 0.2,  $\text{CH}_2\text{Cl}_2$ ). UV ( $\text{CH}_2\text{Cl}_2$ )  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 220 nm. IR (neat)  $\nu_{\text{max}}$ : 3385, 1738  $\text{cm}^{-1}$ .  $^1\text{H}$ - and  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ): see Tables 1 and 2, respectively. HR-ESI-MS  $m/z$  373.1990 ( $[\text{M}+\text{Na}]^+$ , calcd for  $\text{C}_{20}\text{H}_{30}\text{O}_5\text{Na}$ , 373.1991).

Cespitulins C (**3**): light yellow powder.  $[\alpha]_{\text{D}}^{22} +28.3$  (c 0.25,  $\text{CH}_2\text{Cl}_2$ ). UV ( $\text{CH}_2\text{Cl}_2$ )  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 220 nm. IR (film)  $\nu_{\text{max}}$  3444, 1731  $\text{cm}^{-1}$ .  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ): see Tables 1 and 2, respectively. HR-ESI-MS  $m/z$  401.2301 ( $[\text{M}+\text{Na}]^+$ , calcd for  $\text{C}_{22}\text{H}_{34}\text{O}_5\text{Na}$ , 401.2304).

Cespitulin D (**4**): light yellow powder.  $[\alpha]_{\text{D}}^{25} -36.8$  (c 0.25,  $\text{CH}_2\text{Cl}_2$ ); UV ( $\text{CH}_2\text{Cl}_2$ )  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 220 nm. IR (neat)  $\nu_{\text{max}}$  3390, 1757  $\text{cm}^{-1}$ .  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ): see Tables 1 and 2, respectively. HR-ESI-MS  $m/z$  401.2302 ( $[\text{M}+\text{Na}]^+$ , calcd for  $\text{C}_{22}\text{H}_{34}\text{O}_5\text{Na}$ , 401.2304).

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### Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2010.10.057.

### References and notes

- Fabricius, K.; Alderslade, P. *Soft Corals and Sea Fans*; Australian Institute of Marine Science: Townsville MC, 2001. p 146.
- Duh, C.-Y.; El-Gamal, A. A. H.; Wang, S.-K.; Dai, C.-F. *J. Nat. Prod.* **2002**, *65*, 1429.
- Shen, Y.-C.; Cheng, Y.-B.; Kobayashi, J.; Kubota, T.; Takahashi, Y.; Mikami, Y.; Ito, J.; Lin, Y.-S. *J. Nat. Prod.* **2007**, *70*, 1961.
- Shen, Y.-C.; Wu, Y.-R.; Lin, J.-J.; Kuo, Y.-C.; Khalil, A. T. *Tetrahedron* **2007**, *63*, 10914.
- Cheer, C. J.; Smith, D. H.; Djerassi, C.; Tursch, B.; Braekman, J. C.; Daloz, D. *Tetrahedron* **1976**, *32*, 1807.
- Bowden, B. F.; Coll, J. C.; Mitchell, S. J.; Stokie, G. J.; Blount, J. F. *Aust. J. Chem.* **1978**, *31*, 2039.
- Bowden, B. F.; Braekman, J. C.; Mitchell, S. J. *Aust. J. Chem.* **1980**, *33*, 927.
- Shen, Y.-C.; Ho, C.-J.; Kuo, Y.-H.; Lin, Y.-S. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 2369.
- Bowden, B. F.; Coll, J. C.; Tapiolas, D. M. *Aust. J. Chem.* **1983**, *36*, 211.
- Bowden, B. F.; Coll, J. C.; Gulbis, J. M.; Mackay, M. F.; Willis, R. H. *Aust. J. Chem.* **1986**, *39*, 803.
- König, G. M.; Wright, A. D. J. *Nat. Prod.* **1993**, *56*, 2198.
- Shen, Y.-C.; Lin, J.-J.; Wu, Y.-R.; Chang, J.-Y.; Lo, K.-L. *Tetrahedron Lett.* **2006**, *47*, 6651.
- Baloglu, E.; Kingston, D. G. I. *J. Nat. Prod.* **1999**, *62*, 1448.
- Luh, L.-J.; El-Razek, M. H. A.; Liaw, C.-C.; Chen, C.-T. A.; Lin, Y.-S.; Kuo, Y.-H.; Chien, C.-T.; Shen, Y.-C. *Helv. Chim. Acta* **2009**, *92*, 1349.
- Shen, Y.-C.; Lin, Y.-S.; Kuo, Y.-H.; Chen, Y.-B. *Tetrahedron Lett.* **2005**, *46*, 7893.
- Duh, C.-Y.; Li, C.-H.; Wang, S.-K.; Dai, C.-F. *J. Nat. Prod.* **2006**, *69*, 1188.
- Cheng, Y.-B.; Chen, C.-Y.; Kuo, Y.-H.; Shen, Y.-C. *Chem. Biodiversity* **2009**, *6*, 1266.
- Cheng, Y.-B.; Lo, K.-L.; Chen, C.-Y.; Khalil, A. T.; Shen, Y.-C. *Helv. Chim. Acta* **2008**, *91*, 2308.
- Chang, J.-Y.; El-Razek, M. H. A.; Kuo, Y.-H.; Shen, Y.-C. *Helv. Chim. Acta* **2009**, *92*, 2146.
- Shen, Y.-C.; Lo, K.-L.; Kuo, Y.-H.; Kuo, Y.-C.; Chen, C.-H.; Khalil, A. T. *J. Nat. Prod.* **2008**, *71*, 1993.